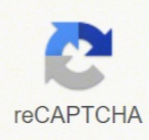




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Next

ACT

ACT Reading:
Top 10 Study Tips

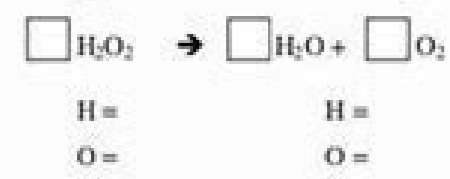
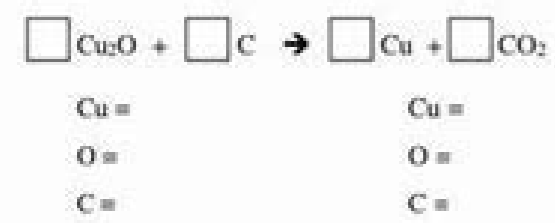
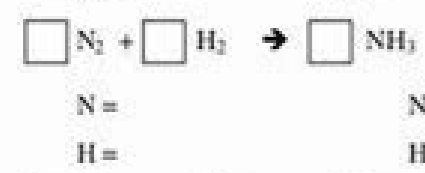
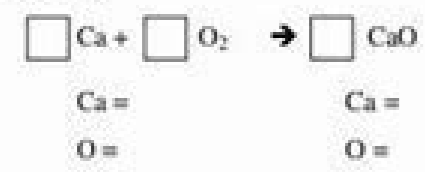
Magosh

Balancing Act

Name _____

Atoms are not _____ or _____ during a chemical reaction. Scientists know that there must be the _____ number of atoms on each _____ of the _____. To balance the chemical equation, you must add _____ in front of the chemical formulas in the equation. You cannot _____ or _____ subscripts!

- 1) Determine number of atoms for each element. $\square \text{Mg} + \square \text{O}_2 \rightarrow \square \text{MgO}$
- 2) Pick an element that is not equal on both sides of the equation. Mg = Mg =
- 3) Add a coefficient in front of the formula with that element and adjust your counts. O = O =
- 4) Continue adding coefficients to get the same number of atoms of each element on each side.

Try these:T. Trimpe 2006 <http://sciencepot.net/>**ANSWER KEY TO TEST 2**

Section 1	Section 2	Section 3	Section 4	Section 5	Section 6	Section 7
1. A	1. C	1. D	1. A	1. A	1. D	1. A
2. E	2. D	2. B	2. D	2. E	2. B	2. B
3. B	3. B	3. D	3. D	3. B	3. C	3. E
4. A	4. C	4. B	4. A	4. C	4. E	4. B
5. A	5. E	5. E	5. A	5. C	5. C	5. A
6. E	6. B	6. E	6. B	6. D	6. C	6. B
7. E	7. E	7. D	7. B	7. A	7. D	7. B
8. D	8. A	8. B	8. C	8. D	8. C	8. C
9. A	9. C	9. A	9. B	9. E	9. E	9. D
10. A	10. A	10. A	10. C	10. C	10. C	10. D
11. A	11. B	11. B	11. A	11. B		11. C
12. A	12. A	12. E	12. B	12. C		12. B
13. D	13. C	13. E	13. A	13. A		13. A
14. B	14. C	14. B	14. D			14. E
15. A	15. B	15. E	15. C			15. B
16. D	16. C	16. E	16. 9			16. A
17. B	17. C	17. C	17. 3/2			17. C
18. A	18. B	18. D	18. 20			18. E
19. D	19. A	19. A	19. 4			19. E
20. E	20. B	20. C	20. - 45/28			20. B
21. C	21. B	21. B	21. 20/3			21. B
22. B	22. C	22. D	22.			22. A
23. E	23. D	23. D	4200/601			23. C
24. D	24. C	24. B	23. 8.66			24. D
25. D	25. D	25. C	24. 30			25. E
26. E		26. E	25. 27/20			
27. C		27. D				
28. E		28. C				
29. A		29. E				
30. D		30. A				
		31. D				
		32. C				
		33. C				
		34. E				
		35. D				

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