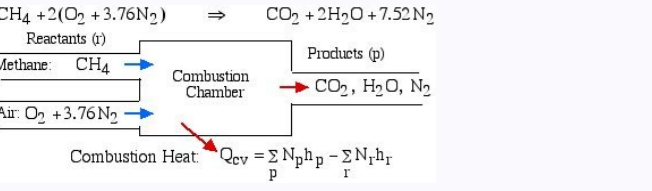
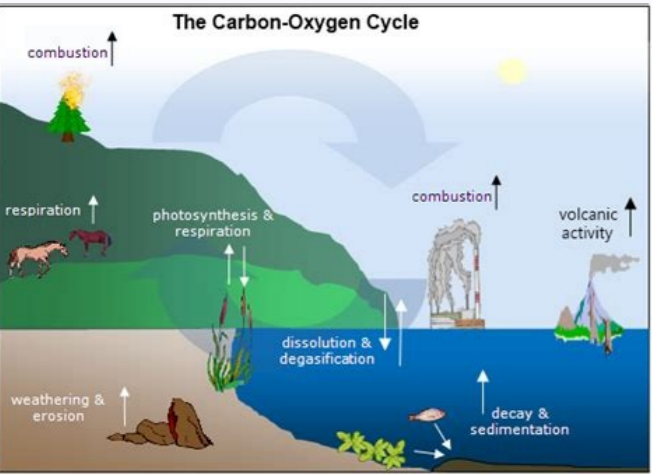
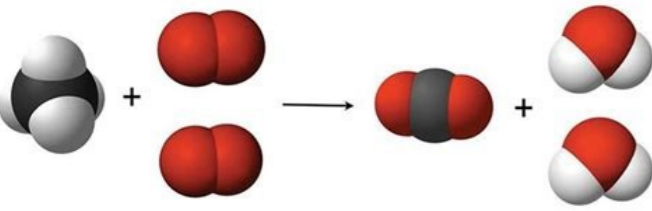


Methane to carbon dioxide equation

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Carbon Cycle - Combustion, Metabolism

$$\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{energy}$$

or oxygen carbon water (heat)

$$\text{C}_6\text{H}_{12}\text{O}_6 \quad \text{dioxide}$$

represents any fossil fuel, food, organic matter

In a policy address last week, President Barack Obama made the reduction of greenhouse gas emissions in the United States a key priority in the nation's fight against climate change. Now, a newly released geological report points to a promising way to cut down on the amount of harmful carbon dioxide pumped into the atmosphere: inject and store it inside rocks deep underground. The U.S. Geological Survey (USGS) conducted a detailed assessment and found 36 regions across the country that have the proper subterranean conditions to store between 2,400 to 3,700 metric gigatons of carbon dioxide underground — a process known as geologic carbon sequestration. One metric gigaton is equal to a billion metric tons. In a separate report released in early June, the U.S. Energy Information Administration, an organization that collects and analyzes statistics on energy production and consumption, projects the United States will emit approximately 5.4 metric gigatons of fossil fuel-related carbon dioxide in 2013, which includes coal, natural gas and petroleum emissions. Based on these estimates, the USGS findings represent a vast, untapped resource that could help reduce carbon dioxide emissions and mitigate the impact they have on Earth's climate, said Briana Mordick, a geologist at the Natural Resources Defense Council (NRDC), a nonprofit environmental advocacy group headquartered in New York City. "This is just one tool in a range of options that we have, but it's an important one to give us additional time to transition from fossil fuels to nonfossil fuel energy," Mordick told LiveScience. (The Reality of Climate Change: 10 Myths Busted) As part of its survey, the USGS excluded areas of the country that are considered freshwater sources, and limited their assessment to rock layers at depths at which the carbon dioxide would be under sufficient pressure to remain in a liquid state, which would help the carbon dioxide mix in with the briny water found underground. The study identified the largest storage potential in the Coastal Plains region, which encompasses much of the Gulf Coast. This area could account for roughly 2,000 metric gigatons, or 65 percent, of the country's storage potential, according to the USGS report. Other areas with considerable storage capacity include the Alaska region and the Rocky Mountains. Going beneath the surface Geologic carbon sequestration involves capturing the exhaust gases from power plants before they are released into the atmosphere, and separating the carbon dioxide from the rest of the emissions. This carbon dioxide is then cooled and compressed into a so-called supercritical state, which means it has properties between a liquid and a gas, Mordick explained. Next, the supercritical carbon dioxide travels through a network of underground pipelines to a site where it is pumped through a well into subsurface rocks. "The idea is that the carbon dioxide will be trapped there pretty much indefinitely," Mordick said. "Things like oil, gas and brine are trapped in the subsurface for millions of years, so basic geologic principles tell us this is possible. In some ways, it's mimicking natural geologic processes." To do this, the carbon dioxide needs to be injected deep underground, between at least 3,000 and 15,000 feet (914 and 4,600 meters), said Peter Warwick, chief of the geologic carbon sequestration project at the USGS, which put out the sequestration report. In addition, certain types of rocks are more suited to hold carbon dioxide. [Video: How Carbon Capture & Sequestration Works] You want a rock that has what we call porosity, which means there are small, open areas within the rock, and permeability, which is the ability for fluid to move through the rock," Warwick said. Sandstone or limestone rock formations are particularly good storage reservoirs, but equally important are the layers of rock over the top that act as a cap, sealing in the carbon dioxide, Mordick said. Without this robust rock layer, carbon dioxide could seep out and leak to the surface, reaching the atmosphere anyway. "There has to be a good ceiling formation above — something like shale, with low porosity and low permeability," Mordick said. "Essentially, it's like a lid on top of the storage formation that prevents carbon dioxide from migrating vertically." Leaks are one of the primary concerns surrounding geologic carbon sequestration, and researchers around the country are assessing the risks involved, which includes studying the types of conditions that could cause carbon dioxide to escape. Trapped underground One possible way the gas could escape is by seeping into a shallower rock formation, where it might then spread and eventually make its way to the surface, said Ronald Falta, a professor in the Department of Environmental Engineering and Earth Sciences at Clemson University in Clemson, S.C. In 2009, Falta and a colleague, Larry Murdoch, received an \$891,000 grant from the Environmental Protection Agency (EPA) to research how to safely store carbon dioxide in geological formations. The project, which also involves Sally Benson, director of Stanford University's Global Climate & Energy Project, is in its final year. Falta said that while leaky carbon dioxide is a major concern, the idea of storing material in subsurface rocks is a well-understood process. "People have been storing natural gas in underground formations for years with very few problems," Falta said. "If these sites are studied carefully, and if they're deep enough, I think the risk is low. But, it's still a major issue that we're going to have to address before anything is done, while the carbon dioxide is being injected, and after it's injected. We need to think: How do we safeguard against leaks, and what are we going to do if it does?" Geologic carbon sequestration is currently regulated by the EPA, under its Class VI rules for injection wells. Under these rules, companies or organizations are required to monitor the site for leaks for at least 50 years after the injection process. Mordick, at the NRDC, said the Class VI guidelines are the most stringent rules the EPA has written, and they are designed to regulate the entire sequestration process, from the selection of the storage site to the decades following. Falta said that over time, different trapping mechanisms will naturally help contain the carbon dioxide (CO₂), but monitoring how the carbon dioxide initially moves through the limestone or sandstone rocks will be critical. "Carbon dioxide dissolves in water under those high pressures, so eventually it's all going to dissolve and not have a tendency to rise," he explained. "Over longer periods, it will turn into minerals and carbonates, so it's mostly in the early periods, when you have a buoyant plume of CO₂, that you have to be really careful. Follow the money The USGS report did not evaluate the economic viability of geologic carbon sequestration, but the cost of deploying these types of capture and storage technologies could be one of the main barriers to actually employing this strategy. For one, extracting carbon dioxide from power plant emissions is a costly process. [Top 10 Craziest Environmental Ideas] It's really expensive to separate the carbon dioxide from the flue gasses coming out of the power plants," Falta said. "That's where the major cost is going to be, and it has been done at small and medium scales, but not at the massive scales that we might be talking about for large power plants." Warwick said the USGS intends to publish a follow-up report on the economics involved with geologic carbon sequestration, based on the results of their initial study. "There is a significant buy-in, so all this development and infrastructure comes with a cost," Warwick said. "If you're willing to pay for the cost to capture CO₂ and put it into the ground, then it could make a significant impact." The USGS is also investigating other risks involved with injecting carbon dioxide deep underground, including whether this process could induce unwanted seismic activity, Warwick added. Injection of waste water from fracking, or hydraulic fracturing, has been linked to increased seismicity in areas where the injection occurs. Still, geologic carbon sequestration represents an enticing way to reduce the nation's amount of greenhouse gas emissions, Falta said, and an opportunity to lessen the environmental impact of coal-fired power plants. "The U.S. has more of these rock formations than any other country, and more than any other continent, so in that respect, we're kind of lucky," Falta said. "It will probably boil down to a question of economics. Will people think it's worth it to do this, or should we continue to use coal? And we have a lot of coal, too." Follow Denise Chow on Twitter @denisechow. Follow LiveScience @livescience, Facebook & Google+. Original article on LiveScience.com. A mild winter, new car efficiency standards and the continued switch from power plants run by coal to those fueled by natural gas, a cleaner-burning fuel, were behind a 3.8-percent drop in U.S. carbon dioxide emissions in 2012, announced by the U.S. Energy Information Administration Monday (Oct. 21). The drop was the second largest since 1990, beat out by the drop of 7.1 percent in 2009, which was attributed in large part to the recession that hit the country that year. Emissions have dropped for five out of the last seven years, the EIA said, and current emissions are down 12 percent from a peak in 2007. In raw numbers, the country released 5,290 million metric tons of carbon dioxide in 2012, down from 5,498 million metric tons released in 2011, according to EIA numbers. The decline came despite a rising population and an increase in gross domestic product, which generally equates to more consumption. "This latest drop in energy-related carbon emissions is reason for cautious optimism that we're already starting to move in the right direction," Michael Mann, a Penn State climate scientist, told the Associated Press. "But this alone will not lead us toward the dramatic carbon reductions necessary to avoid dangerous climate change." The rising global release of carbon dioxide over the past few centuries, largely since the onset of the Industrial Revolution, has steadily increased the proportion of the greenhouse gas in Earth's atmosphere. The latest report from the Intergovernmental Panel on Climate Change, the UN-created body that examines the latest science on the causes and implications of climate change, put the share of carbon dioxide in the atmosphere at 390.5 parts per million in 2011, a 40-percent increase from 1750 and the highest reported levels of carbon dioxide in the last 800,000 years. (Parts per million, or ppm, are used to note the concentration of a gas in the air, or the number of molecules of gas, in this case 390.5, for every million molecules of air.) All of that carbon dioxide increased Earth's average global temperature by 1.6 degrees Fahrenheit (0.89 degrees Celsius) from 1901 to 2012. The IPCC said in this report that it is "extremely likely" that human activities, such as industrial activity, logging and power generation, have caused most of this warming. [Video: Earth's Surface Warming Since 1880] The IPCC and leading climate scientists have said that greenhouse gas emissions, particularly carbon dioxide, will need to be curtailed significantly to avoid much more drastic temperature rises and the knock-on effects they can have on the planet, including changes to ecosystems and animal ranges, changes to weather patterns, warming of the oceans and melting of polar ice. One factor behind 2012 decline in emissions was the continued introduction of more energy-efficient cars and other vehicles. In August of last year, the Obama administration introduced new Environmental Protection Agency fuel standards aimed at increasing fuel economy to 54.5 miles (88 kilometers) per gallon by 2025, in part as an effort to reduce the nation's carbon dioxide emissions. A press release from the Environmental Protection Agency at the time of the announcement revealed the program would reduce emissions by 6 billion metric tons over its lifetime. The continuing switch to power plants fueled by natural gas rather than coal, which emits more carbon dioxide, also helped to lower the nation's emissions in 2012. As did a relatively mild winter and cool summer for much of the country, which reduced the demand for heating fuel and for electricity to power air conditioning, respectively, yesterday's EIA release noted. The six-month total of carbon dioxide emitted so far in 2013 is 2,664 million metric tons, about on par with last year's six-month mark. Follow Andrea Thompson @AndreaTOAP, Pinterest and Google+. Follow us @livescience, Facebook & Google+. Original article on LiveScience.

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